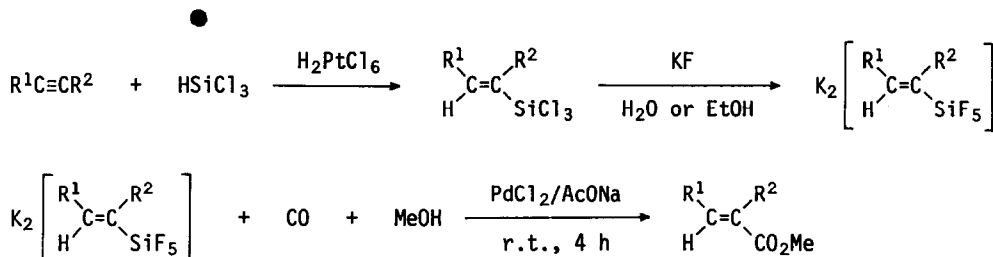


(E)- $\alpha,\beta$ -UNSATURATED CARBOXYLIC ESTERS FROM (E)-ALKENYLPENTAFLUROSILICATES  
BY PALLADIUM-PROMOTED CARBONYLATION<sup>1</sup>

Kohei Tamao, Toshio Kakui, and Makoto Kumada\*

*Department of Synthetic Chemistry, Kyoto University, Yoshida, Kyoto 606, Japan*

Organopentafluorosilicates,  $M_2[RSiF_5]$ , have recently been recognized as versatile intermediates in various organic syntheses.<sup>2</sup> Previously, we have reported that (E)-1,4-dienes can be prepared stereoselectively by the reaction of (E)-alkenylpentafluorosilicates, readily available by hydrosilylation of alkynes, with allylic halides in the presence of catalytic amounts of palladium salts.<sup>2d</sup> As a further example of versatility of alkenylpentafluorosilicates in the carbon-carbon bond formation, we present here the carbonylation of alkenylpentafluorosilicates. This reaction is indeed the first case of carbonylation of organosilicon compounds. Thus, we find that (E)-alkenylpentafluorosilicates readily react with carbon monoxide (atmospheric pressure) in the presence of palladium salts and sodium acetate in methanol to give (E)- $\alpha,\beta$ -unsaturated carboxylic esters in excellent yields.



The most remarkable feature may be the high stereoselectivity of the carbonylation step. The platinum-catalyzed hydrosilylation of 1-alkynes, excepting *t*-butyl- and phenyl-acetylene, forms terminal (E)-1-alkenylsilanes mainly, together with non-terminal adducts as minor products.<sup>3</sup> Since these isomers can be separated by fractional distillation, the present procedure provides

Table I. Carbomethoxylation of  $K_2[RSiF_5]^a$ 

R in $K_2[RSiF_5]$	Product	Yield (%) <sup>b</sup>
$\begin{array}{c} \text{CH}_3(\text{CH}_2)_5 \\ \diagdown \\ \text{C}=\text{C} \\ \diagup \\ \text{H} \end{array}$	$\begin{array}{c} \text{CH}_3(\text{CH}_2)_5 \\ \diagdown \\ \text{C}=\text{C} \\ \diagup \\ \text{H} \end{array} \begin{array}{c} \text{H} \\ \diagdown \\ \text{CO}_2\text{CH}_3 \end{array}$	91 (83) <sup>c</sup>
$\begin{array}{c} (\text{CH}_3)_3\text{C} \\ \diagdown \\ \text{C}=\text{C} \\ \diagup \\ \text{H} \end{array}$	$\begin{array}{c} (\text{CH}_3)_3\text{C} \\ \diagdown \\ \text{C}=\text{C} \\ \diagup \\ \text{H} \end{array} \begin{array}{c} \text{H} \\ \diagdown \\ \text{CO}_2\text{CH}_3 \end{array}$	90
$\begin{array}{c} \text{C}_6\text{H}_5 \\ \diagdown \\ \text{C}=\text{C} \\ \diagup \\ \text{H} \end{array} \begin{array}{c} \text{H}^d \\ \diagdown \\ \end{array}$	$\begin{array}{c} \text{C}_6\text{H}_5 \\ \diagdown \\ \text{C}=\text{C} \\ \diagup \\ \text{H} \end{array} \begin{array}{c} \text{H} \\ \diagdown \\ \text{CO}_2\text{CH}_3 \end{array}$	76
$\begin{array}{c} \text{CH}_3\text{OCH}_2 \\ \diagdown \\ \text{C}=\text{C} \\ \diagup \\ \text{H} \end{array}$	$\begin{array}{c} \text{CH}_3\text{OCH}_2 \\ \diagdown \\ \text{C}=\text{C} \\ \diagup \\ \text{H} \end{array} \begin{array}{c} \text{H} \\ \diagdown \\ \text{CO}_2\text{CH}_3 \end{array}$	61 (57) <sup>c</sup>
$\begin{array}{c} \text{CH}_3\text{O}_2\text{C}(\text{CH}_2)_8 \\ \diagdown \\ \text{C}=\text{C} \\ \diagup \\ \text{H} \end{array}$	$\begin{array}{c} \text{CH}_3\text{O}_2\text{C}(\text{CH}_2)_3 \\ \diagdown \\ \text{C}=\text{C} \\ \diagup \\ \text{H} \end{array} \begin{array}{c} \text{H} \\ \diagdown \\ \text{CO}_2\text{CH}_3 \end{array}$	72
$\begin{array}{c} \text{CH}_3(\text{CH}_2)_3 \\ \diagdown \\ \text{C}=\text{C} \\ \diagup \\ \text{H} \end{array} \begin{array}{c} (\text{CH}_2)_3\text{CH}_3^d \\ \diagdown \\ \end{array}$	$\begin{array}{c} \text{CH}_3(\text{CH}_2)_3 \\ \diagdown \\ \text{C}=\text{C} \\ \diagup \\ \text{H} \end{array} \begin{array}{c} (\text{CH}_2)_3\text{CH}_3 \\ \diagdown \\ \text{CO}_2\text{CH}_3 \end{array}$	88
$\text{C}_6\text{H}_5-$	$\text{C}_6\text{H}_5\text{CO}_2\text{CH}_3$	trace
$\text{C}_6\text{H}_5\text{CH}_2-$	$\text{C}_6\text{H}_5\text{CH}_2\text{CO}_2\text{CH}_3$	trace

<sup>a</sup> Carried out on a 1-mmol scale in the same manner as described in the text, unless otherwise noted. <sup>b</sup> Yields were determined by GLC, based on the silicate. <sup>c</sup> A 3-mmol scale reaction, product being isolated by distillation. <sup>d</sup> These silicates were prepared in EtOH, others in H<sub>2</sub>O.

the regio- and stereo-selective transformation of alkynes into (*E*)- $\alpha,\beta$ -unsaturated carboxylic esters. The use of sodium acetate as a base afforded the most satisfactory results; tertiary amines in place of sodium acetate gave only moderate to poor yields, *e.g.*, yield of methyl 2-nonenolate was 40% with Et<sub>3</sub>N and 20% with pyridine. While PdCl<sub>2</sub> and PdBr<sub>2</sub> exhibited similar reactivities, Pd(OAc)<sub>2</sub> was somewhat less active and no carbonylation occurred with PdCl<sub>2</sub>(PPh<sub>3</sub>)<sub>2</sub>. The carbonylation proceeded smoothly at room temperature (around 25°C), but very slowly at 0°C.

Simplicity and efficiency of the present reaction are demonstrated by the following typical

procedure. To a mixture of  $K_2[(E)-C_6H_{13}CH=CHSiF_5]$  (936 mg; 3.0 mmol), sodium acetate (480 mg; 5.9 mmol) and anhydrous methanol (15 ml), through which had been bubbled carbon monoxide during 5 min, was added palladium chloride (585 mg; 3.3 mmol). The mixture was stirred under the carbon monoxide atmosphere (a CO balloon) at room temperature (ca. 25°C) for 4 h. Almost immediately the color changed from brown to black (possibly palladium black). Ether was added and the mixture filtered. The filtrate was washed three times with 10% NaCl aqueous solution, dried over magnesium sulfate, and distilled (bulb-to-bulb) to give 424 mg (83% yield) of methyl (*E*)-2-nonenolate:  $^1H$  NMR (100 MHz,  $CCl_4$ , TMS)  $\delta$  0.92 (t, 3H), 1.33 (br. 8H), 2.20 (br. q, 2H), 3.67 (s, 3H), 5.76 (double t,  $J = 16Hz, 1Hz, 1H$ ), 6.86 (double t,  $J = 16Hz, 7Hz, 1H$ ); IR, 1730, 1660, 975  $cm^{-1}$ .

Representative results summarized in Table I were obtained under the same conditions as above. The present procedure — a sequence of hydrosilylation of alkynes, silicate formation, and carbonylation — is capable of tolerating some functional groups such as ester and ether. An internal alkenyl-silicate was also carbonylated with equal efficiency, as shown by the preparation of methyl (*E*)-2-butyl-2-heptenoate, the authentic sample of which was prepared by the hydroalumination-carboxylation procedure.<sup>4</sup> Phenyl- and alkyl-silicates, unfortunately, gave only trace amounts of carbonylation products.<sup>5</sup>

Preparation of  $\alpha,\beta$ -unsaturated carboxylic esters via carbonylation of organometallics has so far been known only for alkenylmercurials.<sup>6</sup> While carbonylation of alkenyl halides with nickel carbonyl proceeds with retention of configuration,<sup>7</sup> palladium-catalyzed reactions suffer from rather low stereoselectivity.<sup>8</sup> Our present procedure, therefore, provides a convenient, practically useful, and stereoselective route to (*E*)- $\alpha,\beta$ -unsaturated carboxylic esters from acetylenes.

Although the mechanism has not yet been clarified, the reaction may proceed most probably via the carbon monoxide insertion into an alkenyl-palladium bond<sup>9</sup> resulting from the transfer of the alkenyl group from silicon to palladium(II). Mechanistic study and applications of this carbonylation reaction are now in progress.

*Acknowledgments.* We thank the Grant-in-Aid for Scientific Research of the Ministry of Education (No. 303523), the Yamada Science Foundation, and Shin-etsu Chemical Co., Ltd. for support of the work.

## REFERENCES AND NOTES

1. Organofluorosilicates in Organic Synthesis. 5. Part 4, see ref. 2d.
2. (a) K. Tamao, J. Yoshida, M. Takahashi, H. Yamamoto, T. Kakui, H. Matsumoto, A. Kurita, and M. Kumada, *J. Am. Chem. Soc.*, **100**, 290 (1978); (b) K. Tamao, T. Kakui, and M. Kumada, *ibid.*, **100**, 2268 (1978); (c) J. Yoshida, K. Tamao, A. Kurita, and M. Kumada, *Tetrahedron Lett.*, 1809 (1978); (d) J. Yoshida, K. Tamao, M. Takahashi, and M. Kumada, *ibid.*, 2161 (1978).
3. E. Lukevics, Z. V. Belyakova, M. G. Pomerantseva, and M. G. Voronkov, *J. Organometal. Chem. Library*, **5**, 1 (1977).
4. G. Zweifel and R. B. Steele, *J. Am. Chem. Soc.*, **89**, 2754 (1967).
5. In a control experiment, (*E*)-styryltrimethylsilane gave only 13% yield of methyl (*E*)-cinnamate under similar conditions. *Cf.*, W. P. Weber, R. A. Felix, A. K. Willard, and K. E. Koenig, *Tetrahedron Lett.*, 4701 (1971).
6. R. C. Larock, *J. Org. Chem.*, **40**, 3237 (1975).
7. E. J. Corey and L. S. Hegedus, *J. Am. Chem. Soc.*, **91**, 1233 (1969).
8. A. Schoenberg, I. Bartoletti, and R. F. Heck, *J. Org. Chem.*, **39**, 3318 (1974).
9. P. M. Maitlis, *The Organic Chemistry of Palladium*, Vol. 1 and 2, Academic Press, New York, 1971.

(Received in Japan 15 November 1978)